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Binaphthyl scaffolded peptoids via ring-closing metathesis reactions and their anti-bacterial activities

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ABSTRACT

An efficient synthesis of four new acyclic and four new cyclic binaphthyl-based cationic peptoids is described. These compounds had anti-bacterial activities with MIC values of $4-62~\mu g/mL$ against Staphylococcus aureus.

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With the increasing spread of antibacterial resistance, 1-3 including resistance by pathogenic bacteria to vancomycin, ^{4,5} there is a strong imperative for new antibacterials.^{6,7} In this context, we have undertaken a programme investigating the design and synthesis of cyclic cationic peptoids linked by a hydrophobic scaffold as potential antibacterial agents, and thus far, have shown the binaphthyl (1)⁸ and carbazole (2) scaffolds^{9,10} within these cyclic peptoids produce antibacterial agents, whilst the smaller indole and phenyl based cyclic peptoids $(3)^{11}$ and $(4)^{12}$ had weak activities or were inactive (Fig. 1). Therefore, as part of this programme targeting new peptoid derivatives as antibacterial agents we investigated the preparation of further novel binaphthyl cyclic cationic peptides. 11a.b and their acyclic analogues 9a.b. In this study the peptide moieties are attached to the 2 and 2' positions of the binaphthyl system, rather than the 3 and 3' positions as in peptoid 1. This modification made these new derivatives synthetically more accessible than our earlier compounds (compound 1 was obtained in twelve synthetic steps) and avoided the problems associated with the extra, uncontrolled stereogenic centre in the peptoid side chain.8 Furthermore, these target compounds maintained the key structural units that we have found that are important for antibacterial activity, (1) a basic amino acid residue for potential electrostatic interaction with the carboxy group of terminal D-Ala or D-

Lac; (2) a peptoid moiety for potential H-bonding interactions with both $\,$ D-Ala or $\,$ D-Lac; and (3) a hydrophobic 1,1'-binaphthyl system.

These target peptoids were prepared in four and five steps, respectively from the known and racemic mono-O-allylbiphathyl 5¹³ as shown in Scheme 1. The phenolic group of 5 was O-alkylated under basic conditions with ethyl bromoacetate and ethyl 4-bromobutyrate to give the esters **6a** and **6b**, respectively. Saponification of these esters with aqueous lithium hydroxide/THF and then acidification of the final reaction mixture resulted in production of their respective carboxylic acids, 7a and 7b (Scheme 1). A peptide coupling reaction of these carboxylic acids with p-lys- N^{ϵ} -Boc-L-allylglycine¹⁴ under standard conditions¹² gave the dipeptoids 8a and 8b, respectively in good yields (85-88%). N-Boc deprotection of 8a and 8b by exposure to TFA, followed by anion exchange with HCl, provided the hydrochloride salts 9a and 9b, respectively (Scheme 1).12 Treatment of 8a and 8b with 10 mol % Grubbs' first generation ruthenium catalyst [(Cy₃P)₂Ru[CHPh]Cl₂] resulted in successful ring-closing metathesis reactions and gave the cyclic peptoids 10a and 10b, respectively in good to excellent yields (77–93%, the E/Z ratios could not be readily determined due to peak overlap) (Scheme 1). N-Boc deprotection of these compounds by treatment with TFA, 15 followed by anion exchange with HCl, realised the hydrochloride salts 10a and 10b, respectively (Scheme 1).15,16

Compounds **9a,b** and **11a,b** underwent N-guanidination reactions by treatment with N,N'-bis(tert-butoxycarbonyl)-N''-triflylguanidine to give their respective homoarginine analogues **12a,b**

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$$\begin{array}{c} \text{CH}_3\\ \text{NH}_3^+\\ \text{OH}\\ \text{O$$

Figure 1. MIC values against S. aureus of vancomycin and cyclic peptoids 1-4.

and **14a,b**, respectively (Schemes 2 and 3).¹⁶ These compounds were then converted to their hydrochloride salts **13a,b** and **15a,b**, as shown in Schemes 2 and 3.

To assist with SAR studies on the above set of deprotected compounds the mono-peptoid **17** was prepared as shown in Scheme 4. A coupling reaction of the carboxylic acid **7a** with D-lys-N^E-Boc(OMe) under standard peptide conditions using DCC and HOBt in MeCN¹² gave the peptoid **16** in 71% yield. N-Boc deprotection of **16** with TFA, followed by anion exchange with HCl provided the hydrochloride salt **17** (Scheme 4).

The synthesized cationic peptoids **9a**, **9b**, **11a**, **11b**, **13a**, **13b**, **15a**, **15b** and **17** were tested against the Gram-positive bacterium Staphylococcus aureus (ATCC6538) and the results are shown in Figure 2. The positive control, vancomycin, showed a MIC value of 1.95 μ g/mL. The acyclic peptoid **9a** showed the highest antibacterial activity with a MIC value of 4 μ g/mL. This compound was more active than its cyclic counterpart **11a** which had a MIC value of 15–31 μ g/mL. Their three carbon chain elongated and ring enlarged analogues **9b** and **11b**, respectively, were significantly less active. Surprisingly, the acyclic and cyclic peptoids having a homoarginine side chain, **13a**, **13b**, **15a** and **15b**, all had MIC values of 8 μ g/mL. Thus, except for **9a**, all homoarginine derivatives were

Scheme 1.

9b (n = 3)

more active than their lysine counterparts. This suggested that the more basic and multi-N-centred guanidine residue was responsible for this enhanced activity. In contrast, the mono-peptoid **17** was not as active (MIC of 32 μg/mL) as the related dipeptoids **9a** and **9b** suggesting that the C-terminal allylglycinate residue also contributes to antibacterial activity. Svendsen¹⁷ has provided a pharmacophore model for peptoid antibacterial compounds which indicates that two hydrophobic and two cationic sites are important for antibacterial activity. In our case the C-terminal allylglycinate residue of **9a**, **9b**, **13a** and **13b** would represent the second, albeit considerably smaller, hydrophobic group when compared to the binaphthyl residue. These results highlight the importance of hydrophobicity within the peptoid for antibacterial activity and indicate that in this series cyclic and acyclic peptoids have

Scheme 2.

Scheme 3.

Scheme 4.

	n	AA	MIC μg/mL
11a	1	Lys	15-31
11b	3	Lys	62
15a	1	homo-Arg 8	
15b	3	homo-Arg 8	

Figure 2. MIC values against *S. aureus* of acyclic and cyclic peptoids as their hydrochloride salts.

similar anti-bacterial activities. While the configuration of the binaphthyl unit in the tested compounds was *R,S* we have found in earlier studies that the configuration of the binaphthyl unit had little impact on antibacterial.¹⁸

In conclusion, the acyclic cationic peptide **9a** can be readily prepared in four synthetic steps and has greater antibacterial activity against *S. aureus* than the structurally more complex and synthetically more challenging cyclic peptoid **1**. The study thus provides a platform for further development of antimicrobial agents with improved activity against *S. aureus*. We are currently extending these studies to molecules that incorporate two hydrophobic residues and two cationic residues and these studies will be reported in the future publications.

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References and notes

- 1. Nordmann, P.; Naas, T.; Fortineau, N.; Poirel, L. Curr. Opin. Microbiol. 2007, 10, 436.
- 2. Nicolaou, K. C.; Boddy, C. N. C. Sci. Am. 2001, 284, 54.
 - Noble, W. C.; Virani, Z.; Cree, R. G. A. FEMS Microbiol. Lett. 1992, 93, 195.
- 4. Kahne, D.; Leimkuhler, C.; Lu, W.; Walsh, C. Chem. Rev. 2005, 105, 425.

- 5. Appelbaum, P. C. Clin. Microbiol. Infect. 2006, 12, 16.
- 6. Rice, L. B. Am. J. Med. **2006**, 119, S11.
- 7. Wright, G. D.; Sutherland, A. D. Trends Mol. Med. 2007, 13, 260.
- 8. Bremner, J. B.; Coates, J. A.; Coghlan, D. R.; David, D. M.; Keller, P. A.; Pyne, S. G. *New J. Chem.* **2002**, *26*, 1549.
- 9. Bremner, J. B.; Coates, J. A.; Keller, P. A.; Pyne, S. G.; Witchard, H. M. Synlett 2002, 219.
- Bremner, J. B.; Coates, J. A.; Keller, P. A.; Pyne, S. G.; Witchard, H. M. *Tetrahedron* 2003, 59, 8741.
- Au, V. S.; Bremner, J. B.; Coates, J. A.; Keller, P. A.; Pyne, S. G. Tetrahedron 2006, 62, 9373.
- Boyle, T. P.; Bremner, J. B.; Coates, J. A.; Deadman, J.; Keller, P. A.; Pyne, S. G.; Rhodes, D. *Tetrahedron* 2008, 64, 11270.
- 13. Nakamura, Y.; Hollenstein, R.; Zsindely, J.; Schmid, H.; Oberhaensli, W. E. *Helv. Chim. Acta* **1975**, *58*, 1949.

- Boyle, T. P.; Bremner, J. B.; Coates, J. A.; Deadman, J.; Keller, P. A.; Pyne, S. G.; Somphol, K. Eur. J. Med. Chem. 2009, 44, 1001.
- Boyle, T. P.; Bremner, J. B.; Coates, J. A.; Keller, P. A.; Pyne, S. G. Tetrahedron 2005, 61, 7271.
- 16. Yields refer to material that had been purified by column chromatography. All compounds (unless specified as mixtures of diastereomers) were homogeneous by TLC analysis and judged to be of >95% purity based upon ¹H NMR analysis. Structures were confirmed by NMR and MS analysis.
- For work on other cationic peptides and a pharmocophore model see: (a) Strom, M. B.; Haug, B. E.; Skar, M. L.; Stensen, W.; Stiberg, T.; Svendsen, J. S. J. Med. Chem. 2003, 46, 1567; (b) Haug, B. E.; Stensen, W.; Stiberg, T.; Svendsen, J. S. J. Med. Chem. 2004, 47, 4159; (c) Haug, B. E.; Stensen, W.; Svendsen, J. S. Biorg. Med. Chem. Lett. 2007, 17, 2361.
- 18. Unpublished results from our laboratory.